A least-squares plane $[3.494 x-2.667 y+6.700 z$ $=2 \cdot 1565]$ through the six atoms $\mathrm{O}(1), \mathrm{O}(2), \mathrm{C}(3)$, $C(4), C(5)$ and $O(3)$ reveals that the $\alpha, \beta$-unsaturated carboxylic acid is essentially planar. Two features of this structure are somewhat curious: the $C(7)$ methyl carbon is nearly in the plane $(-0.0002 \AA)$ described above and also the double bond between $C(2)$ and $C(3)$ is cis to the shorter $\mathrm{C}-\mathrm{O}$ carboxylic acid bond, i.e. $\mathrm{C}(1)-\mathrm{O}(1)$.

The carboxyl end of this structure is close to a crystallographic inversion center so that a hydrogenbonded dimer is a more accurate representation of the solid.

All $\mathrm{C}-\mathrm{Cl}$ bond lengths are equivalent to one another within experimental error and these lengths agree well with what is expected (Kennard, 1962). From the final residual electron density map it is clear that there is some disorder in the crystal due to different rotamers of the trichloromethyl group. We believe that this fact accounts for the slightly high final $w R$ and residual electron density.

A view of the packing is given in Fig. 2. Intermolecular hydrogen bonds holding two asymmetric units together are shown with dashed lines. No other unusual short intermolecular contacts were noted.

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Fig. 2. A view of the packing.

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# Structure of 5-Fluoroarabinosylcytosine 

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#### Abstract

C}_{9} \mathrm{H}_{12} \mathrm{FN}_{3} \mathrm{O}_{5}, M_{r}=261 \cdot 2\), orthorhombic, $P 2_{1} 2_{1} 2_{1}, \quad a=5.927$ (2),$\quad b=7.882$ (1),$\quad c=$ 23.069 (4) $\AA, \quad U=1077.7 \AA^{3}, \quad Z=4, \quad D_{x}=$ $1.61 \mathrm{~g} \mathrm{~cm}^{-3}$, Мо $K \alpha, \lambda=0.71073 \AA, \mu=1.35 \mathrm{~cm}^{-1}$, $F(000)=544, T=296 \mathrm{~K}, R=0.030$ for 1077 unique observed reflections. The $N$-glycosidic torsion angle, $\chi$,


is $164.0(2)^{\circ}$, in the anti range; the sugar pucker is ${ }^{2} E$, $P=165.4(2)^{\circ}, \psi_{m}=37.3(2)^{\circ}$; the $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ conformation is $a p$ with $\gamma=173.8(2)^{\circ}$. These are typical values for arabinosyl nucleosides. The $a p$ conformation about the $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ bond precludes the formation of an intramolecular hydrogen bond between $\mathrm{O}\left(5^{\prime}\right)$ and
$O\left(2^{\prime}\right)$ as is found in some other arabinosyl nucleosides. The absolute configuration of the title compound is not known.

Introduction. The arabinosyl nucleosides possess a wide range of antiviral and other biological properties (Saenger, 1984). One of the most potent of such compounds is arabinosylcytosine (ara-C). We report here the structure of the 5 -fluoro derivative of ara-C, designated ara-FC. This determination is one of our series of nucleoside structure determinations.

Experimental. Needle-shaped crystals grown from aqueous solution. Accurate cell dimensions and crystal orientation matrix determined on a CAD-4 diffractometer by least-squares treatment of 25 reffections in the range $8^{\circ}<\theta<12^{\circ}$. Crystal dimensions $0.08 \times$ $0.10 \times 0.20 \mathrm{~mm}$; intensities of reflections with indices $h$ 0 to $7, k 0$ to $10, l 0$ to 29 with $2^{\circ}<2 \theta<54^{\circ}$ measured; $\omega-2 \theta$ scans; graphite-monochromatized Mo $K \alpha$ radiation; intensities of three reflections chosen as standards monitored every 2 h and showed no evidence of crystal decay; 1460 reflections measured, 1383 unique; 1077 had $I>3 \sigma(I)$ and used in the structure solution and refinement; $R_{\mathrm{int}}=0.045$. Data corrected for Lorentz and polarization effects but not for absorption. Phase problem solved using direct methods. An $E$ map revealed all non-hydrogen atoms; H atoms located and their positions and isotropic temperature factors refined; all other atoms refined anisotropically. The final cycle of refinement included 212 variable parameters, $R=0.030$ and $w R=0.038$, $w=1 /\left(\sigma^{2} F_{o}+0.05 F_{o}^{2}\right)$. Max. $\Delta / \sigma<0.00 ;$ max. difference-map peak 0.23 e $\AA^{-3}$, min. difference-map peak $-0.43 \mathrm{e} \dot{A}^{-3}$. Anomalous-dispersion effects were included in $F$. Scattering factors and anomalousdispersion corrections from International Tables for $X$-ray Crystallography (1974). All calculations were performed on a PDP11/73 computer using SDP-Plus (Frenz, 1978). Conformation parameters are described according to the IUPAC-IUB Joint Commission on Biochemical Nomenclature (1983).

Discussion. The atomic numbering used is shown in the perspective drawing (Fig. 1); atomic parameters, bond distances and bond angles are listed in Tables 1 and 2.* The bond parameters in ara-FC show no unusual features, with, as expected, the $\mathrm{C}\left(1^{\prime}\right)-\mathrm{O}\left(4^{\prime}\right)$ distance $[1.418(3) \AA]$ shorter than the $C\left(4^{\prime}\right)-O\left(4^{\prime}\right)$ distance [ 1.451 (3) $\AA$ ]. The smallest ring angle in the sugar is at

[^0]

Fig. 1. An ORTEP drawing (Johnson, 1965) of the ara-FC molecule with $50 \%$ thermal ellipsoids, showing the atomic numbering.

Table 1. Positional parameters and equivalent isotropic temperature factors $\left(\AA^{2}\right)$ with e.s.d.'s in parentheses

| $B_{\mathrm{eq}}=\frac{1}{3} \breve{-i}^{1} \triangle_{j i j} a_{i}^{*} a_{j}^{*} \mathrm{a}_{i} . \mathrm{a}_{j}$. |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $x$ | $y$ | $z$ | $B_{\text {eq }}\left(\AA^{2}\right)$ |
| F | -0.1824 (3) | 0.5202 (2) | 0.74378 (7) | 3.87 (3) |
| N(1) | $0 \cdot 1508$ (3) | 0.3555 (2) | 0.86023 (8) | 1.81 (4) |
| C(2) | 0.0328 (4) | 0.2083 (3) | 0.87370 (9) | 1.98 (4) |
| N(3) | -0.1557 (3) | $0 \cdot 1696$ (3) | 0.84234 (8) | 2.09 (4) |
| C(4) | -0.2245 (4) | 0.2689 (3) | 0.7993 (1) | 1.93 (4) |
| C(5) | -0.1047 (4) | 0.4192 (3) | 0.7868 (1) | $2 \cdot 23$ (5) |
| C(6) | 0.0787 (4) | 0.4607 (3) | 0.8169 (1) | $2 \cdot 21$ (4) |
| O(2) | $0 \cdot 1018$ (3) | 0.1166 (2) | 0.91298 (8) | $3 \cdot 14$ (4) |
| N(4) | -0.4061 (4) | 0.2252 (3) | 0.76866 (9) | $2 \cdot 60$ (4) |
| C(1') | 0.3597 (4) | 0.3927 (3) | 0.8929 (1) | 1.81 (4) |
| $\mathrm{O}\left(4^{\prime}\right)$ | 0.4803 (3) | 0.5184 (2) | 0.86178 (7) | $2 \cdot 12$ (3) |
| $\mathrm{C}\left(4^{\prime}\right)$ | 0.5899 (4) | 0.6343 (3) | 0.90171 (9) | 1.87 (4) |
| C( $3^{\prime}$ ) | 0.5378 (4) | 0.5660 (3) | 0.96261 (9) | 1.86 (4) |
| C(2') | 0.3184 (4) | 0.4663 (3) | 0.95352 (9) | 1.77 (4) |
| $\mathrm{O}\left(3^{\prime}\right)$ | 0.7163 (3) | 0.4531 (2) | 0.97792 (7) | $2 \cdot 56$ (3) |
| $\mathrm{C}\left(5^{\prime}\right)$ | $0 \cdot 5094$ (6) | 0.8143 (3) | 0.8910 (1) | 3.05 (6) |
| O(5') | 0.5869 (4) | 0.8717 (2) | 0.83587 (8) | 3.63 (4) |
| $\mathrm{O}\left(2^{\prime}\right)$ | $0 \cdot 1377$ (3) | $0 \cdot 5828$ (2) | 0.95349 (8) | $2 \cdot 55$ (3) |

Table 2. Bond lengths $(\AA)$ and bond angles $\left({ }^{\circ}\right)$ with e.s.d.'s in parentheses

| $\mathrm{F}-\mathrm{C}(5)$ | 1.353 (3) | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{O}\left(4^{\prime}\right)$ | 1.418 (3) |
| :---: | :---: | :---: | :---: |
| $N(1)-C(2)$ | 1.390 (3) | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)$ | 1.533 (3) |
| $N(1)-C(6)$ | 1.366 (3) | $\mathrm{O}\left(4^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)$ | 1.451 (3) |
| $\mathrm{N}(1)-\mathrm{C}\left(1^{\prime}\right)$ | 1.479 (3) | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | 1.536 (3) |
| $\mathrm{C}(2)-\mathrm{N}(3)$ | 1.366 (3) | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | 1.518 (4) |
| $\mathrm{C}(2)-\mathrm{O}(2)$ | 1.229 (3) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)$ | 1.534 (3) |
| N(3)-C(4) | $1 \cdot 328$ (3) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right)$ | 1.427 (3) |
| C(4)-C(5) | 1.411 (3) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{O}\left(2^{\prime}\right)$ | 1.411 (3) |
| $\mathrm{C}(4)-\mathrm{N}(4)$ | 1.333 (3) | $\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(5^{\prime}\right)$ | 1.425 (4) |
| C(5)-C(6) | 1.331 (4) |  |  |
| $\mathrm{C}(2)-\mathrm{N}(1)-\mathrm{C}(6)$ | $120 \cdot 8$ (2) | $\mathrm{N}(1)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{O}\left(4^{\prime}\right)$ | 107.6 (2) |
| $\mathrm{C}(2)-\mathrm{N}(1)-\mathrm{C}\left(1^{\prime}\right)$ | 118.2 (2) | $\mathrm{N}(1)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)$ | 114.0 (2) |
| $\mathrm{C}(6)-\mathrm{N}(1)-\mathrm{C}\left(1^{\prime}\right)$ | 121.0 (2) | $\mathrm{O}\left(4^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)$ | $106 \cdot 2$ (2) |
| $\mathrm{N}(1)-\mathrm{C}(2)-\mathrm{N}(3)$ | 118.7 (2) | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{O}\left(4^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)$ | $110 \cdot 1$ (2) |
| $\mathrm{N}(1)-\mathrm{C}(2)-\mathrm{O}(2)$ | 119.2 (2) | $\mathrm{O}\left(4^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | 105.7 (2) |
| $\mathrm{N}(3)-\mathrm{C}(2)-\mathrm{O}(2)$ | 122.1 (2) | $\mathrm{O}\left(4^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | $110 \cdot 2$ (2) |
| $\mathrm{C}(2)-\mathrm{N}(3)-\mathrm{C}(4)$ | 121.0 (2) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | 114.5 (2) |
| $\mathrm{N}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 119.6 (2) | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)$ | 103.0 (2) |
| $\mathrm{N}(3)-\mathrm{C}(4)-\mathrm{N}(4)$ | 119.5 (2) | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right)$ | 107.2 (2) |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{N}(4)$ | 121.0 (3) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right)$ | $110 \cdot 1$ (2) |
| $\mathrm{F}-\mathrm{C}(5)-\mathrm{C}(4)$ | 118.2 (2) | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | 100.5 (2) |
| $\mathrm{F}-\mathrm{C}(5)-\mathrm{C}(6)$ | 121.1 (2) | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{O}\left(2^{\prime}\right)$ | 111.5 (2) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | 120.7 (2) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{O}\left(2^{\prime}\right)$ | 108.1 (2) |
| $\mathrm{N}(1)-\mathrm{C}(6)-\mathrm{C}(5)$ | 119.2 (3) | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(5^{\prime}\right)$ | 109.9 (3) |



Fig. 2. The arrangement of molecules in the unit cell, viewed along a, showing hydrogen bonding.
$\mathrm{C}\left(2^{\prime}\right) \quad\left[\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right) \quad 100 \cdot 5(2)^{\circ}\right]$. This agrees with the observation made by Sundaralingam (1975), to the effect that in the arabinosides the puckered atoms always assume the smallest ring valency angle'. The $N$-glycosidic torsion angle $\chi\left[\mathrm{C}(2)-\mathrm{N}(1)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{O}\left(4^{\prime}\right)\right]$ is $164.0(2)^{\circ}$, in the usual anti range. The sugar pucker, as defined by the pseudorotational phase angles $P=165.4(2)^{\circ}, \quad \psi_{m}=37.3(2)^{\circ}, \quad$ is ${ }^{2} E, \quad \mathrm{C}\left(2^{\prime}\right)$-endo, which is in the same conformational region as all arabinosides studied to date.

The conformation about $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ as defined by the angle $\gamma\left[\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(5^{\prime}\right) 173.8(2)^{\circ}\right]$, is $a p$. In the structures of arabinosyl nucleosides an intramolecular hydrogen bond is sometimes formed between $C\left(2^{\prime}\right)$ and $O\left(5^{\prime}\right)$. Formation of this intramolecular hydrogen bond in ara- FC is precluded by the ap conformation of the $C\left(4^{\prime}\right)-C\left(5^{\prime}\right)$ bond.

The possible intermolecular hydrogen-bond contacts are given in Table 3 (deposited). All possible donor H atoms are involved in these contacts. There is one bifurcated (three-centred) contact involving $\mathrm{H}(\mathrm{N} 4)$ and F and $\mathrm{O}\left(5^{\prime}\right)$ (Fig. 2).

As has been noted in previous halogenated nucleosides (Wilson, Low \& Young, 1983), the halogen atom stacks above a symmetry-related base. The perpendicular distance of the F atom above the mean plane of the base ring atoms is 3.015 (4) $\AA$.

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# Structure of 5-Methylthio-1,3,4-thiadiazole-2-thione* 

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Abstract. $\mathrm{C}_{3} \mathrm{H}_{4} \mathrm{~N}_{2} \mathrm{~S}_{3}, M_{r}=164 \cdot 3$, monoclinic, $P 2_{1} / n$, $a=4 \cdot 126(1), \quad b=9.507(1), \quad c=16.701$ (2) $\AA, \quad \beta=$

[^1]$92 \cdot 12(1)^{\circ}, \quad V=654.7 \AA^{3}, \quad Z=4, \quad D_{m}=1.68, \quad D_{x}=$ $1.67 \mathrm{Mg} \mathrm{m}^{-3}$, Мо $K \alpha, \lambda=0.7107 \AA, \mu=0.98 \mathrm{~mm}^{-1}$, $F(000)=336 \cdot 0, T=293 \mathrm{~K}, R=0.037$ for $900 \mathrm{ob}-$ served reflections. The thiadiazole ring is planar with $S-C(2)$ and $C(2)-N(3)$ partial double bonds. The © 1986 International Union of Crystallography


[^0]:    * Lists of structure amplitudes, anisotropic thermal parameters, H -atom parameters, bond lengths and angles involving H , and details of hydrogen bonding have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 42709 ( 21 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CHI 2HU, England.

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